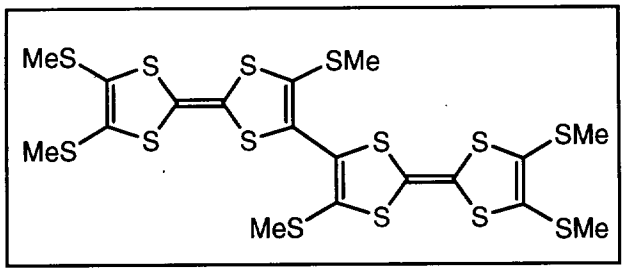
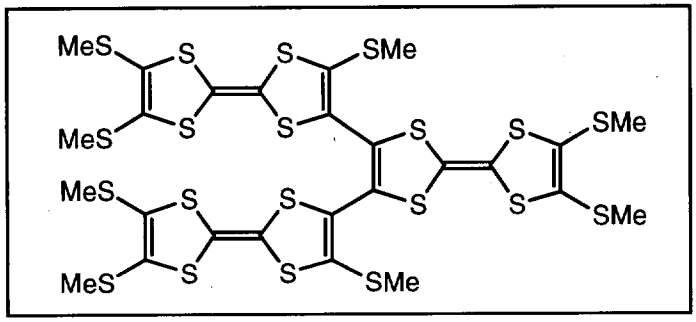
The cyclic voltammograms of compound **2-4**.

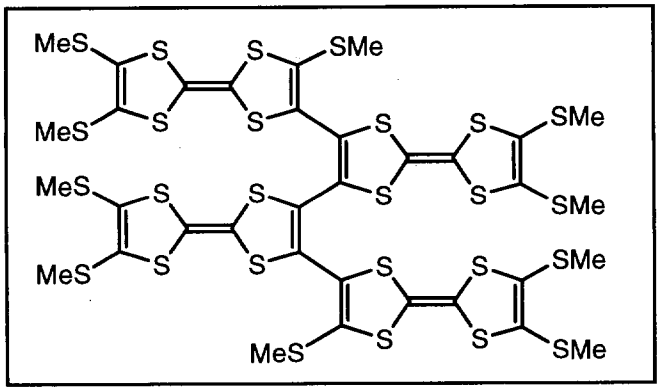


2 $E^1_{1/2} = 0.58$ $E^2_{1/2} = 0.85$



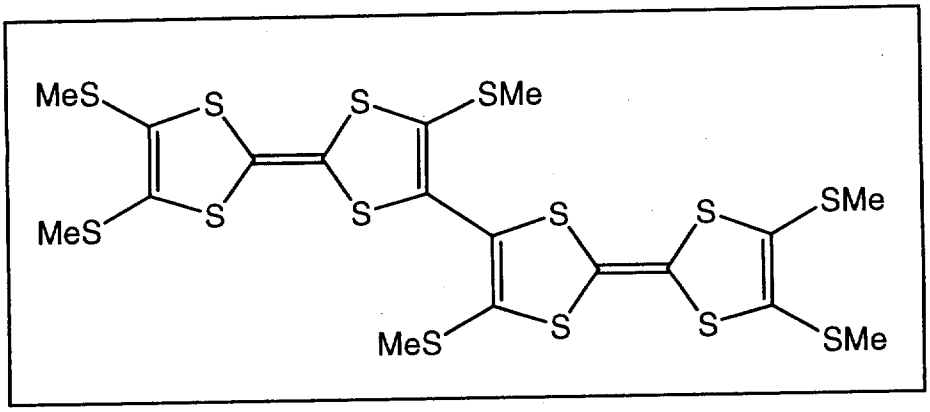
3

$E^1_{1/2} = 0.58$ $E^2_{1/2} = 0.61$ $E^3_{1/2} = 0.82$ $E^4_{1/2} = 0.86$

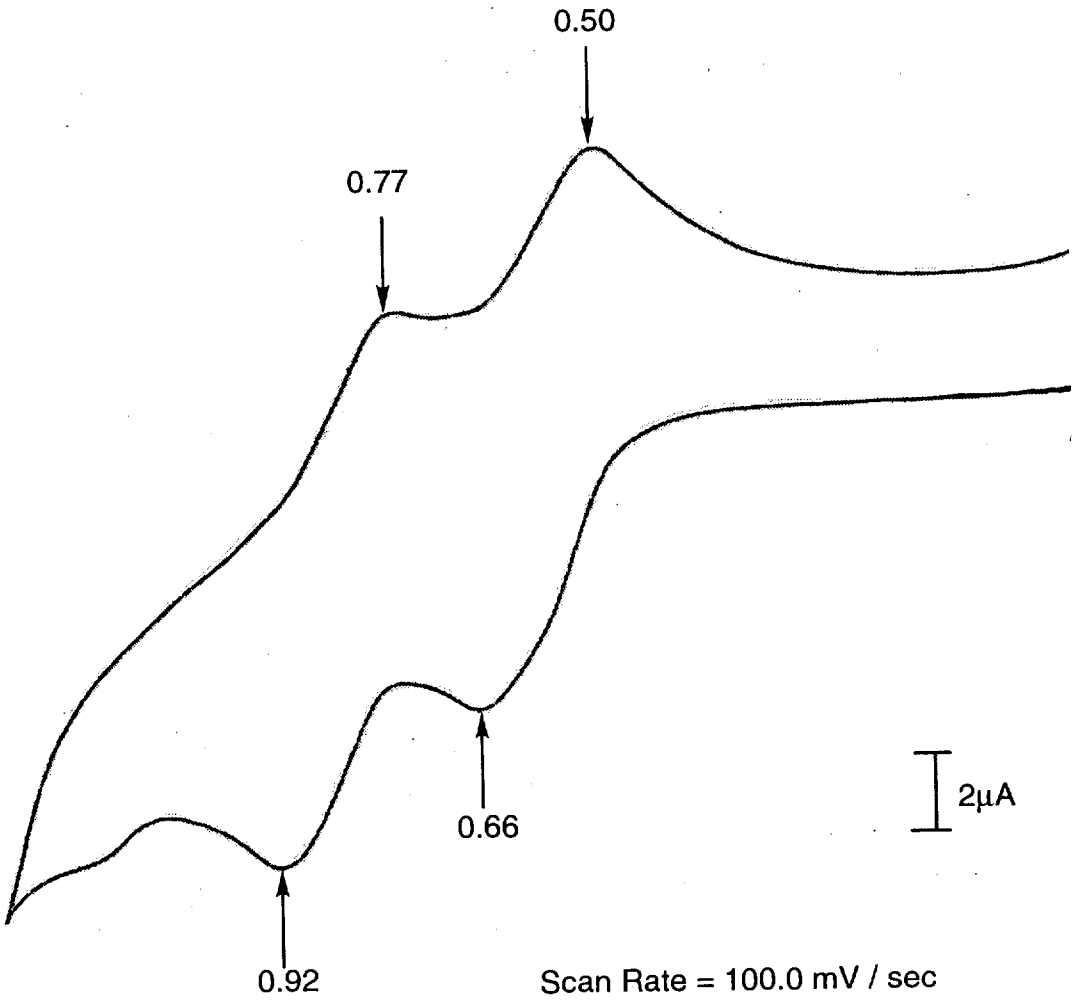


4

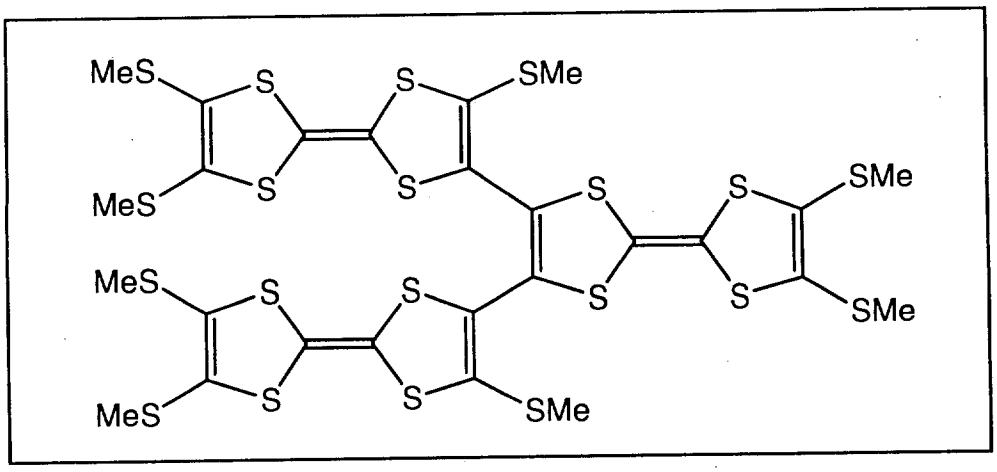
$E^1_{1/2} = 0.51$ $E^2_{1/2} = 0.53$ $E^3_{1/2} = 0.66$ $E^4_{1/2} = 0.88$ $E^5_{1/2} = 0.92$



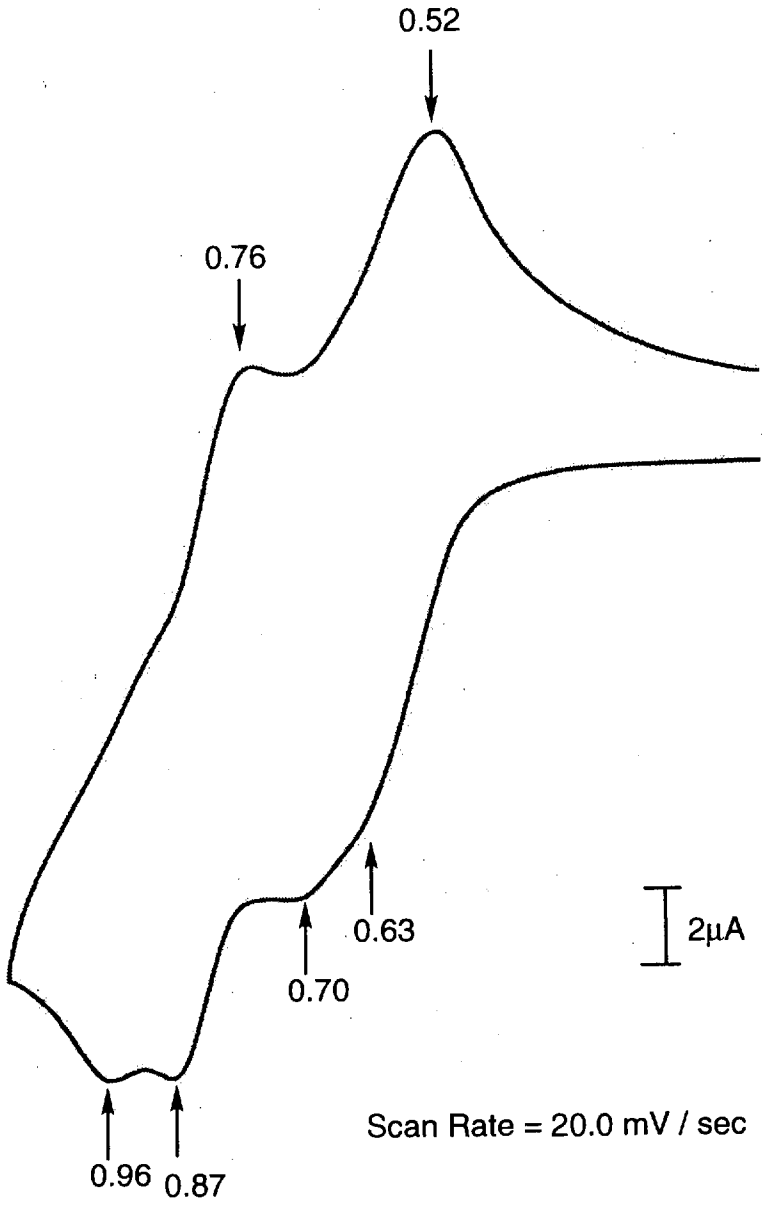
$E^1_{1/2} = 0.58$ $E^2_{1/2} = 0.85$



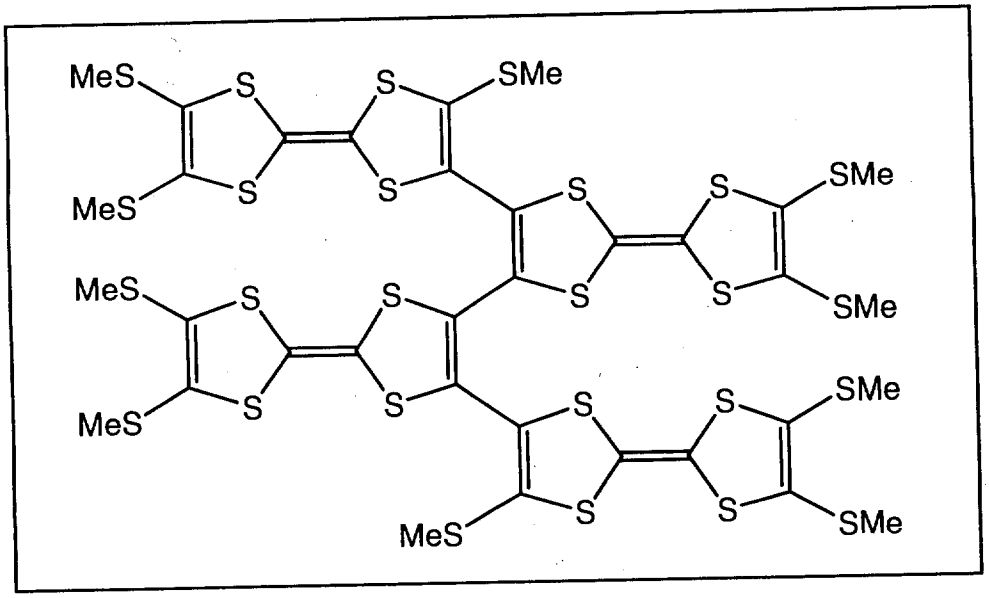
Conditions: Bun4NClO4, benzonitrile, room temperature, Pt working and counter electrodes. Potentials were measured against an Ag/Ag+ electrode and converted to the value *vs* SCE (Fc/Fc+ = 0.46 V)



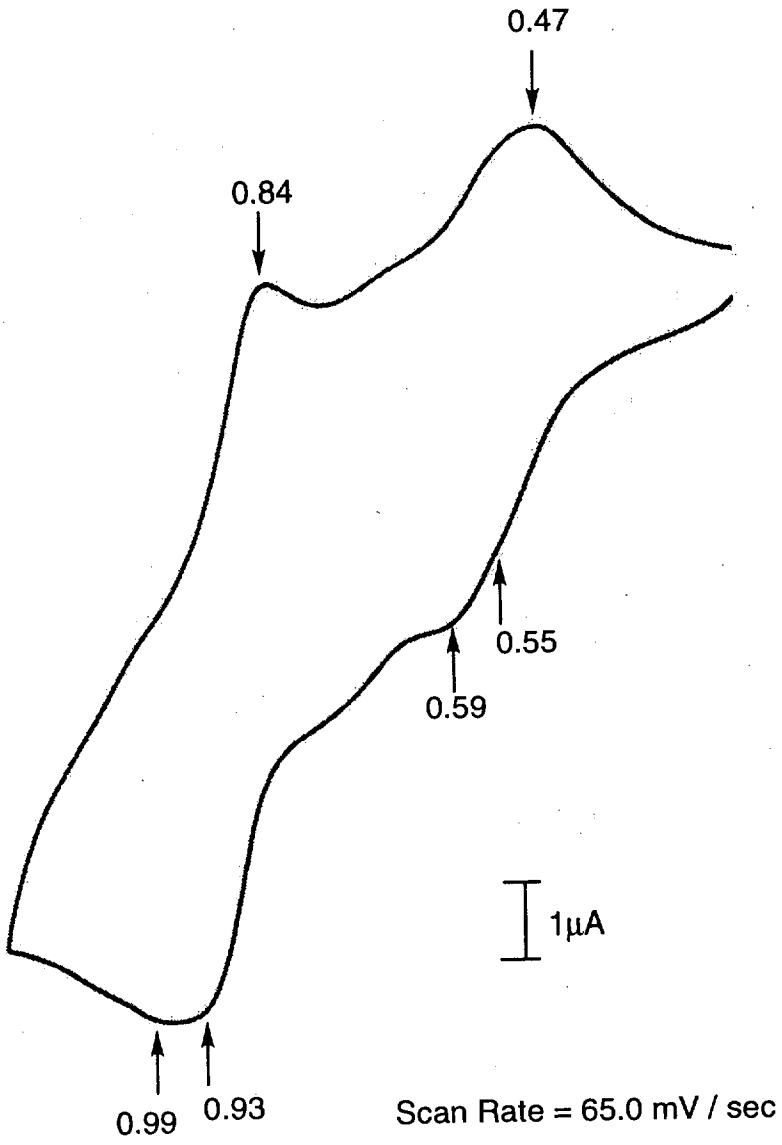
$E^1_{1/2} = 0.58$ $E^2_{1/2} = 0.61$ $E^3_{1/2} = 0.82$ $E^4_{1/2} = 0.86$



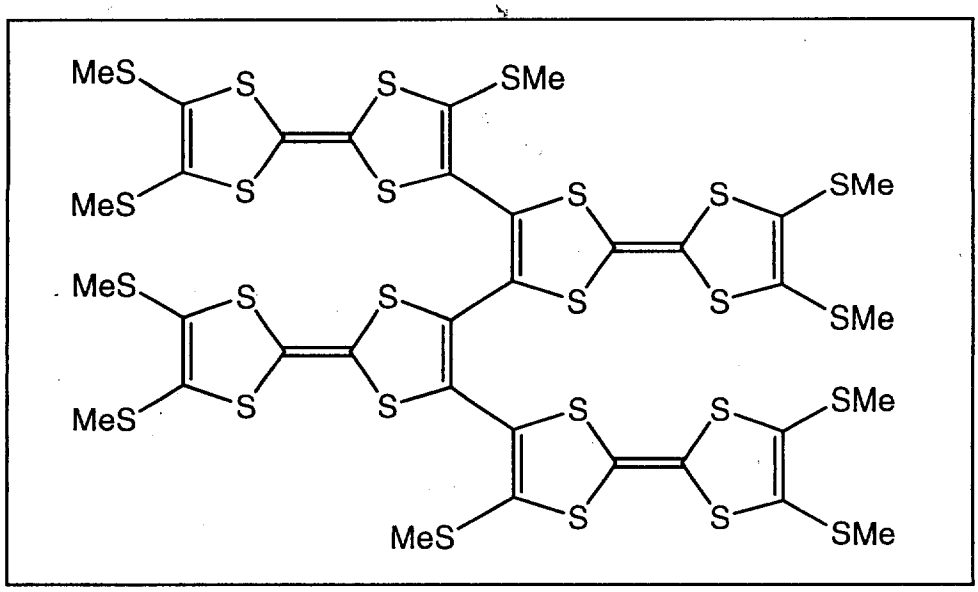
Conditions: Bun4NClO4, benzonitrile, room temperature, Pt working and counter electrodes. Potentials were measured against an Ag/Ag+ electrode and converted to the value *vs* SCE (Fc/Fc+ = 0.46 V)



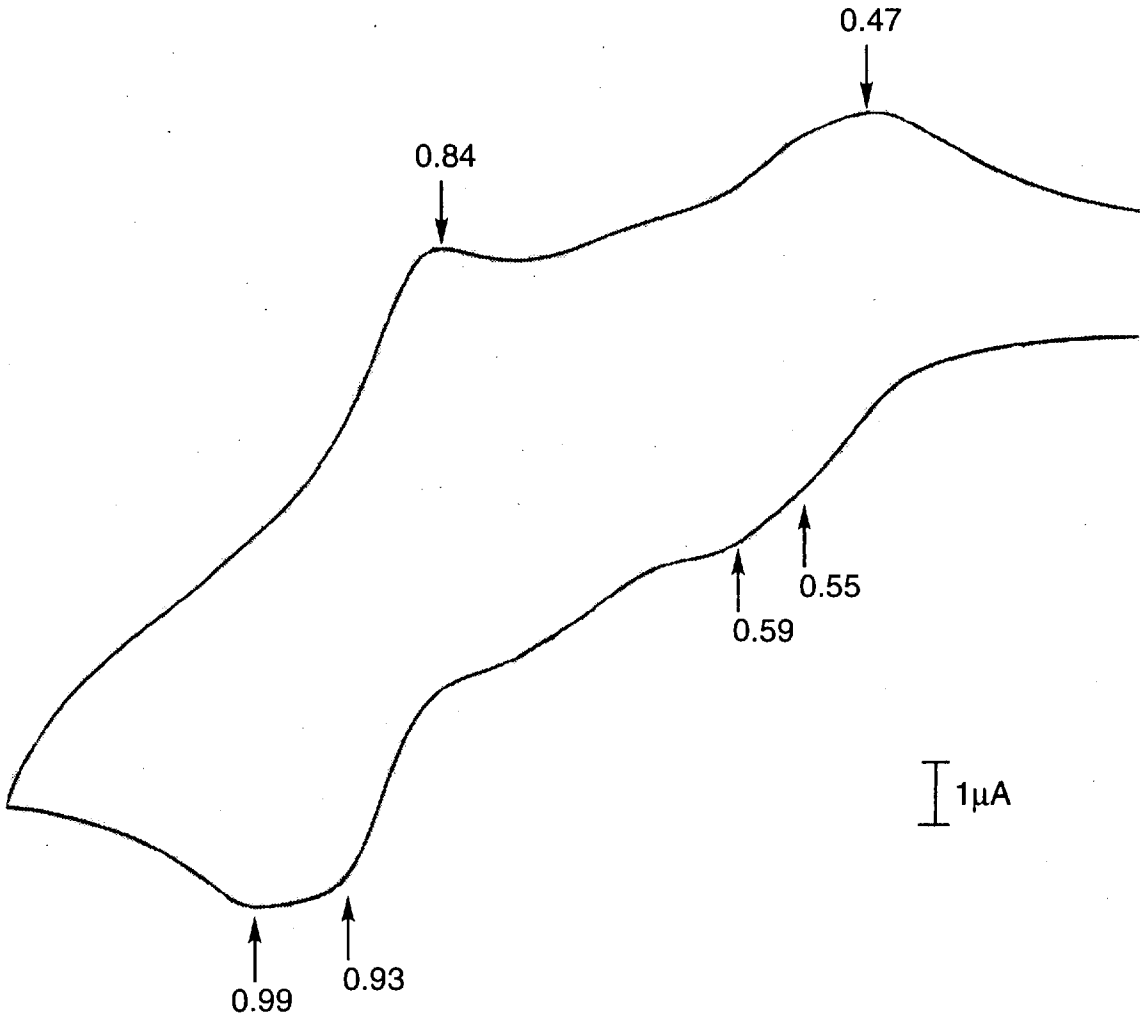
$E^1_{1/2} = 0.51$ $E^2_{1/2} = 0.53$ $E^3_{1/2} = 0.66$ $E^4_{1/2} = 0.88$ $E^5_{1/2} = 0.92$



Conditions: Bun4NClO4, benzonitrile, room temperature, Pt working and counter electrodes. Potentials were measured against an Ag/Ag+ electrode and converted to the value *vs* SCE (Fc/Fc+ = 0.46 V)



$E^1_{1/2} = 0.51$ $E^2_{1/2} = 0.53$ $E^3_{1/2} = 0.66$ $E^4_{1/2} = 0.88$ $E^5_{1/2} = 0.92$



Conditions: Bun4NClO4, benzonitrile, room temperature, Pt working and counter electrodes. Potentials were measured against an Ag/Ag+ electrode and converted to the value *vs* SCE (Fc/Fc+ = 0.46 V)